

# Stereoselective formation of quaternary carbon centres with chiral 3-sulfonyl-1,3-oxazolidines and titanium enolates

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Received 5 July 2001; revised 29 October 2001; accepted 9 November 2001

**Abstract**—The reaction of chiral 2-alkoxy-3-sulfonyl-1,3-oxazolidines and trichlorotitanium enolates was applied for the stereoselective construction of quaternary  $\alpha$ -carbonyl stereocentres on cycloalkanones. The influence of different chiral 1,3-oxazolidines on the selectivity and yield of this reaction has been studied. 2-Alkoxy-3-sulfonyl-1,3-oxazolidines bearing one small group in the 4 position or two *trans*-arranged groups in the 4 and 5 position have been found to give the best results. © 2002 Elsevier Science Ltd. All rights reserved.

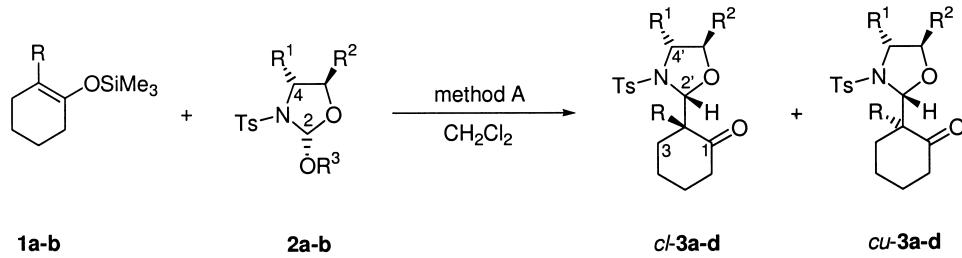
## 1. Introduction

The Lewis acid mediated asymmetric formylation of prochiral silyl enol ethers (e.g. **1**) with optically pure 2-alkoxy-3-sulfonyl-1,3-oxazolidines (e.g. **2**) has been elaborated by our group<sup>1</sup> and the group of C. Scolastico<sup>2</sup> since the early 1990s (Scheme 1, method A). The Lewis acid leads to the formation of a 1,3-oxazolidine cation,<sup>2d</sup> followed by the reaction with the nucleophilic silyl enol ether. The advantages of this reaction have been shown in many synthetic applications.<sup>3</sup> Beneficial properties of the obtained products (e.g. **3**,<sup>‡</sup> Table 1, entries 1 and 2) are the easy separation of diastereomers and compatibility and stability under many reaction conditions. The deprotection of the introduced formyl group can be accomplished by electrolysis,<sup>3e</sup> acetal exchange reactions<sup>2b,3a</sup> and subsequent hydrolysis or other transformations.<sup>3f</sup> During our

synthesis of metachromin A<sup>3k</sup> we were challenged that the construction of quaternary carbon centres gives only poor selectivities<sup>2b</sup> (Table 1, entries 3–5). The previous conversion of the silyl enol ether **1b**<sup>4a</sup> into a titanium enolate led to much better stereoselection and allowed us the preparation of metachromin A in high overall yield. In this article we report our results dealing with this new method in detail.

## 2. Results and discussion

As shown in Scheme 2 we converted the silyl enol ether **1b** into the titanium enolates **4a**<sup>5</sup> and **4b**<sup>6</sup> (methods B and C); both species reacted subsequently with the cation of 2-ethoxy-1,3-oxazolidine **2c**. Due to the Lewis-acidic properties of enolate **4b** no addition of a Lewis acid<sup>7</sup> for



for R, R<sup>1</sup>, R<sup>2</sup> and R<sup>3</sup> see Table 1

**Scheme 1.**

**Keywords:** oxazolidines; enolates; titanium and compounds; asymmetric reactions.

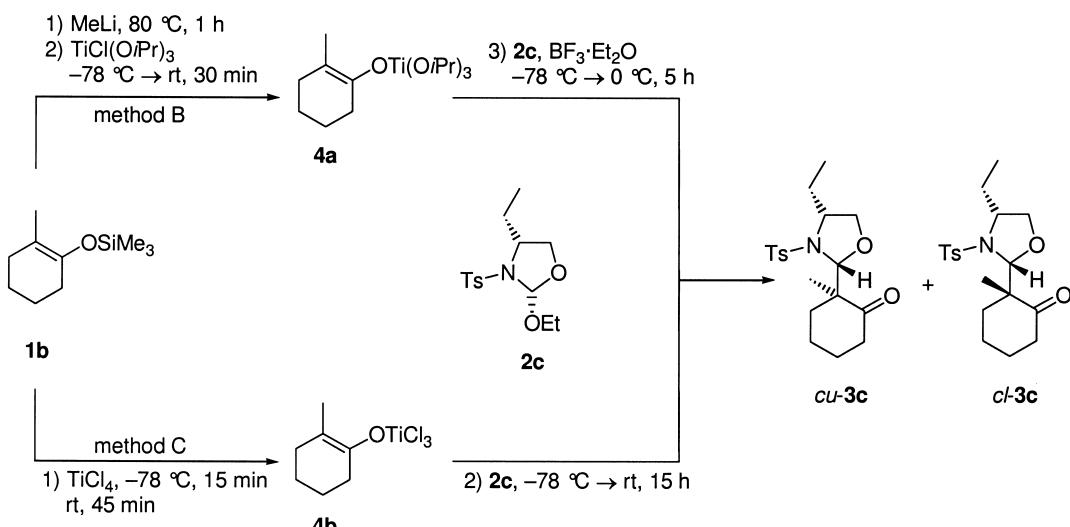
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<sup>†</sup> X-Ray structure analyses.

<sup>‡</sup> Nomenclature: the four possible diastereomers of this reaction are marked as *cl*, *cu*, *tl* and *tu*. *cis* or *trans* (*c* or *t*) indicates the relative configuration at the 1,3-oxazolidine ring (substituents in 2 and 4 position); *like* or *unlike* (*l* or *u*) stands for the relative configuration of the stereocentre in the 2 position of the 1,3-oxazolidine ring and its neighbouring stereogenic carbon centre.

**Table 1.** Reactions of silyl enol ethers **1** and 2-alkoxy-3-sulfonyl-1,3-oxazolidines **2**

Entry	R	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	Educts	Lewis-acid	Temperature (°C)	Product	Yield (%) <sup>a</sup>	dr (cl/cu) <sup>b</sup>
1	H	Et	H	Me	<b>1a+2a</b>	ZnCl <sub>2</sub> ·Et <sub>2</sub> O	0	<b>3a</b> <sup>3c</sup>	75	84:16
2	H	Me	Ph	Me	<b>1a+2b</b>	ZnCl <sub>2</sub> ·Et <sub>2</sub> O	0	<b>3b</b> <sup>1b</sup>	91	>95:5
3	Me	Et	H	Et	<b>1b+2c</b>	ZnCl <sub>2</sub> ·Et <sub>2</sub> O	0	<b>3c</b> <sup>1j</sup>	83	51:49
4	Me	Me	Ph	Me	<b>1b+2b</b>	ZnCl <sub>2</sub> ·Et <sub>2</sub> O	0	<b>3d</b> <sup>1b</sup>	78	50:50
5	Me	Et	H	Me	<b>1b+2a</b>	TiCl <sub>4</sub>	-78	<b>3e</b> <sup>1e</sup>	— <sup>c</sup>	50:50

<sup>a</sup> All yields were determined after flash column chromatography.<sup>b</sup> All diastereomeric ratios were determined by <sup>1</sup>H NMR spectroscopy of the crude products.<sup>c</sup> Not determined.**Scheme 2.**

the generation of the 1,3-oxazolidine cation was necessary in that case. Method B provided the product **3c** with the best selectivity (dr=85:15, Table 2, entry 1) but only with a moderate yield (51%). Because of the much better yield (79%) and the easier practical performance, method C (dichloromethane as the solvent, Table 2, entry 3) was superior.

**Table 2.** Reactions of titanium enolates **4** and 2-ethoxy-3-sulfonyl-1,3-oxazolidine **2c**

Entry	Method	Enolate	Solvent	Yield (%) <sup>a</sup>	dr (cu/cl) <sup>b</sup>
1	B	<b>4a</b>	PhMe	51	85:15
2	C	<b>4b</b>	PhMe	44	78:22
3	C	<b>4b</b>	CH <sub>2</sub> Cl <sub>2</sub>	79	81:19

<sup>a</sup> All yields were determined after flash column chromatography.<sup>b</sup> All diastereomeric ratios were determined by <sup>1</sup>H NMR spectroscopy of the crude products.

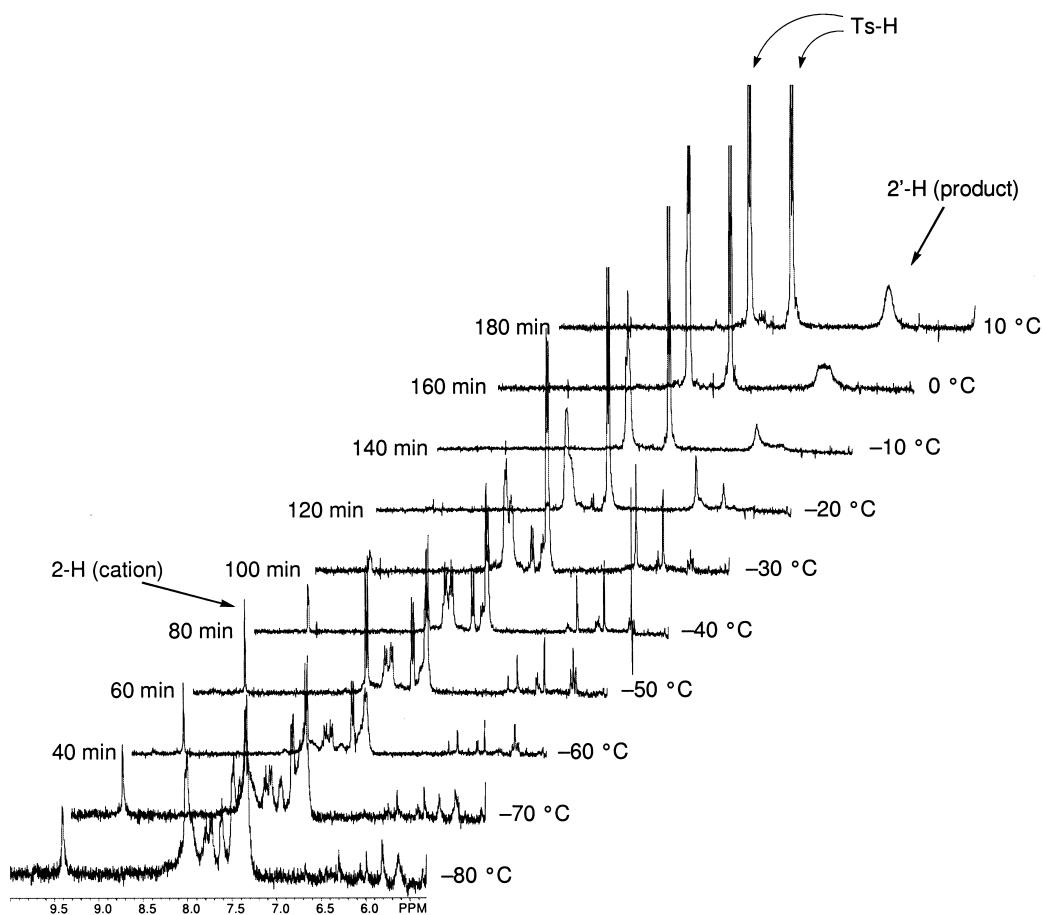
It is noteworthy that the same reagents were used earlier in reverse order;<sup>1e</sup> only a poor selectivity was observed that case (Table 1, entry 5). This indicates that the previous formation of a titanium enolate is essential. This kind of reaction has not been studied before.

To get a further insight into this reaction (Table 2, entry 3) temperature-dependent NMR experiments were done. The reaction was carried out in a carefully sealed NMR tube and the temperature was slowly increased from -80 to 10°C. <sup>1</sup>H

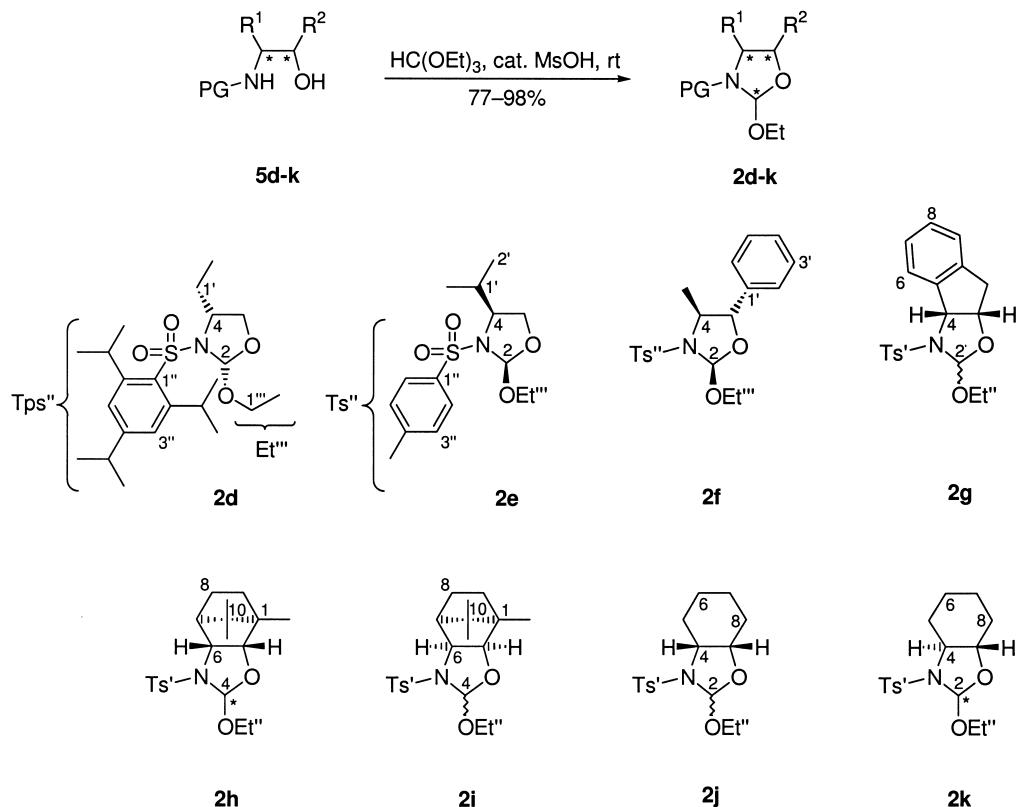
NMR spectra were recorded periodically and are depicted in Fig. 1. A signal at 9.41 ppm (2-H) appearing right from the beginning of the reaction indicates the presence of a 1,3-oxazolidine cation,<sup>2d</sup> formed by the abstraction of the ethoxy group of **2c** by the Lewis-acidic titanium enolate **4b**. Then, at higher temperatures (~−40°C) this signal disappears slowly whereas the signal of the formed product **cu-3c** at 6.28 ppm (2'-H) increases.<sup>§</sup> This signal of the 1,3-oxazolidine **cu-3c** is splitted below a temperature of 0°C (probably due to hindered bond rotations) and shifted down-field in comparison with the same signal (5.36 ppm) recorded in a solvent free of Lewis acids. We conclude that the abstraction of the ethoxy group at low temperature is the first step of the reaction; at higher temperatures the formed cation reacts with the enolate to give the product **3c**. This result reveals that the configuration of the C-2 in reagent **2c** will have no influence on the diastereomeric ratio of the formed 1,3-oxazolidine **3c**.

The variation of the chiral 2-ethoxy-3-sulfonyl-1,3-oxazolidine and its influence on the yield and diastereoselectivity of the studied reaction was of particular interest. Scheme 3 and Table 3 show the results of the preparation of eight new 2-ethoxy-1,3-oxazolidines **2d–k** from known chiral sulfonamides **5d–k**. Especially reagents such as **2g–k** bearing a bicyclic moiety have not been

<sup>§</sup> Some product was formed initially because of imperfect cooling during the placement of the NMR tube into the Bruker AM 360 instrument.



**Figure 1.**  $^1\text{H}$  NMR-spectroscopic study of the reaction of the titanium enolate **4b** and the 2-ethoxy-1,3-oxazolidine **2c**.



**Scheme 3.**

**Table 3.** Preparation of 2-ethoxy-3-sulfonyl-1,3-oxazolidines 2

Entry	Educt	Product	Yield (%) <sup>a</sup>	dr <sup>b</sup>
1	<b>5d</b> <sup>3j</sup>	<b>2d</b> <sup>c</sup>	92	90:10
2	<b>5e</b> <sup>3c</sup>	<b>2e</b>	94	92:8
3	<b>5f</b> <sup>1b</sup>	<b>2f</b>	98	>97:3
4	<b>5g</b> <sup>8</sup>	<b>2g</b>	99	71:29 <sup>d</sup>
5	<b>5h</b> <sup>9</sup> + <b>5i</b> (82:18) <sup>e</sup>	<b>2h</b> + <b>2i</b> (83:17) <sup>f,g</sup>	94	99:1
6	<b>5i</b> <sup>9</sup> + <b>5h</b> (92:8) <sup>e</sup>	<b>2i</b> + <b>2h</b> (96:4)	90	86:14
7	<b>5j</b> <sup>10</sup>	<b>2j</b>	77	54:46 <sup>d</sup>
8	<b>5k</b> <sup>11</sup>	<b>2k</b>	90	73:27 <sup>h</sup>

<sup>a</sup> All yields were determined after flash column chromatography.

<sup>b</sup> All diastereomeric ratios (*cis/trans*) were determined by <sup>1</sup>H NMR spectroscopy of the crude products.

<sup>c</sup> Contained 8% of triethyl orthoformate.

<sup>d</sup> The *cis*- and *trans*-isomer were separated by flash column chromatography.

<sup>e</sup> Sulfonamides **5h** and **5i** were synthesized as an inseparable mixture.

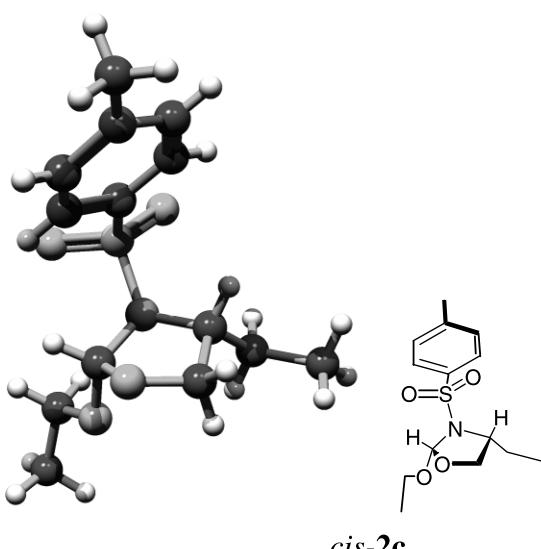
<sup>f</sup> Contained 12% of triethyl orthoformate.

<sup>g</sup> 2-Ethoxy-1,3-oxazolidines **2h** and **2i** were separated by crystallization.

<sup>h</sup> After flash column chromatography only the main diastereomer was isolated in 90% yield.

studied before. All compounds **2** were prepared in good to excellent yields and varying ratios of C-2 epimers. Interestingly, 2-ethoxy-1,3-oxazolidine **2k** seems to epimerize on silica gel and only traces of the minor diastereomer which was detected in the crude product were found after chromatography. In two cases (Table 3, entries 4 and 7) it was possible to separate the C-2 epimers by flash column chromatography. This allowed us to verify our hypothesis that the configuration in this position should have no effect on the diastereoselectivity in the reaction with trichlorotitanium enolates.

The structure of the main diastereomer of reagent **2c** was elucidated by X-ray analysis<sup>12</sup> (Fig. 2). As expected, the relative configuration of the substituents in the 2 and 4 position was proven to be *cis*. Table 4 shows selected NMR signals of the *cis*- and *trans*-isomers of the 2-ethoxy-1,3-oxazolidines **2c–k**. With the exception of **2i** and **2j** the signals (2-H, C-2) of all major diastereomers are shifted more downfield than the signals of the corre-



**Figure 2.** Molecular structure of 2-ethoxy-3-sulfonyl-1,3-oxazolidine *cis*-**2c**.<sup>12</sup>

**Table 4.** Selected NMR data of the 2-ethoxy-1,3-oxazolidines **2**

Entry	Compound	Major diastereomer		Minor diastereomer	
		2-H	C-2	2-H	C-2
1	<b>2c</b>	<b>6.02<sup>a</sup></b>	<b>108.0</b>	5.87	107.1
2	<b>2d</b>	6.02	107.7	5.78	106.7
3	<b>2e</b>	6.02	108.2	5.91	107.5
4	<b>2f</b>	6.22	107.6		
5	<b>2g</b>	6.18	108.8	5.81	107.4
6	<b>2h</b>	5.85 <sup>b</sup>	108.2 <sup>c</sup>	5.63 <sup>b</sup>	— <sup>d</sup>
7	<b>2i</b>	6.08 <sup>b</sup>	111.2 <sup>c</sup>	6.13 <sup>b</sup>	112.3 <sup>c</sup>
8	<b>2j</b>	5.89	105.8	5.97	107.7
9	<b>2k</b>	5.85	107.1	5.02	— <sup>d</sup>

<sup>a</sup> Bold shifts indicate that the configuration was determined by X-ray analyses.

<sup>b</sup> 4-H is the corresponding proton.

<sup>c</sup> C-4 is the corresponding carbon atom.

<sup>d</sup> Not detected.

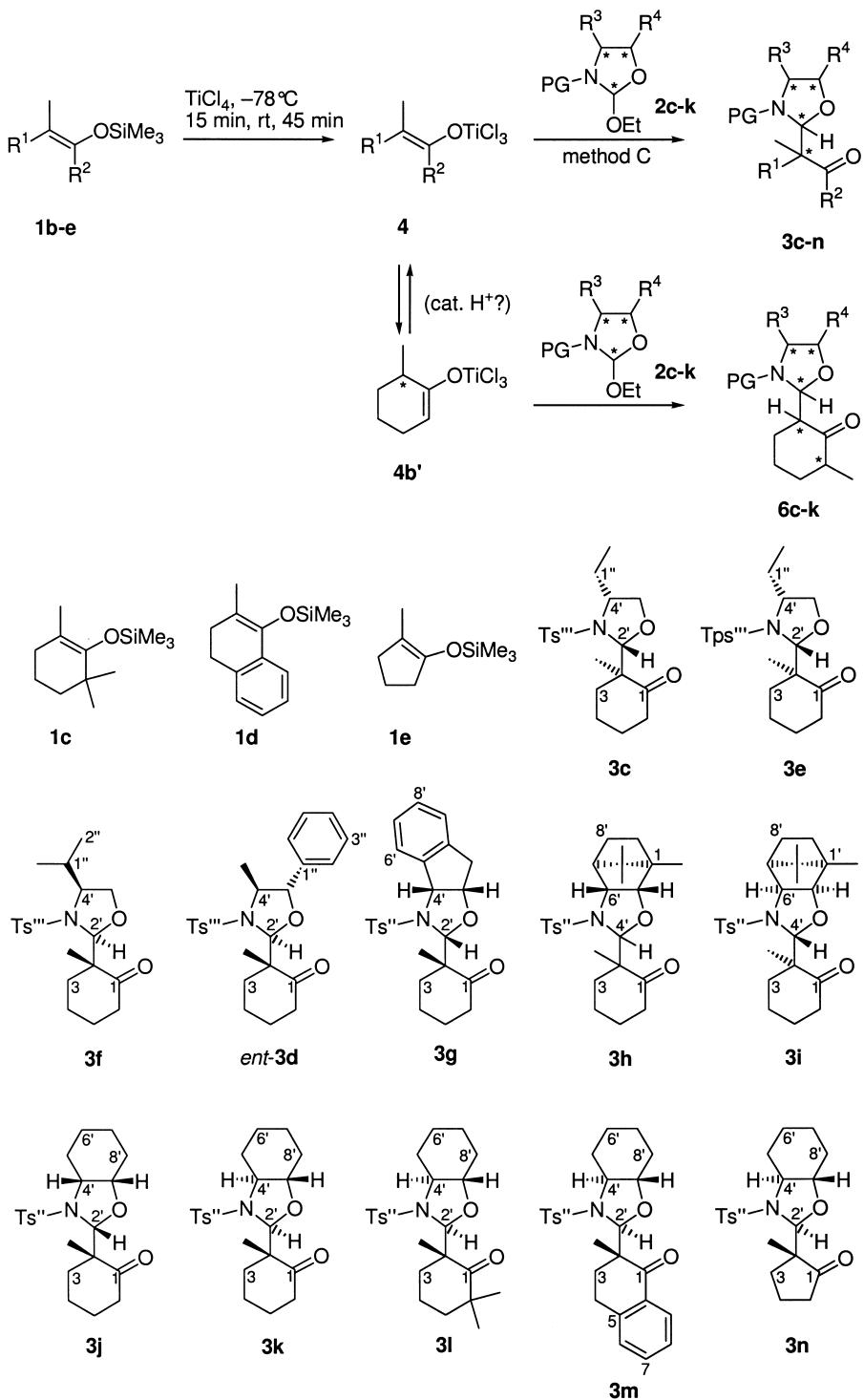
sponding minor isomers. This result may be an indication that in most cases the *cis*-isomers are favoured; only a strong steric shielding of one side of the 1,3-oxazolidine ring in compounds like **2i** or **2j** may lead to a preference for the *trans*-configuration.

The results of the reactions of the titanium enolate **4b** and the 2-ethoxy-1,3-oxazolidines **2c–k** are presented in Scheme 4 and Table 5 (entries 1–11). In most cases the main product showed *cis*-*unlike*-configuration (Table 5, entries 1–4 and 11). In contrast to this, reagent **2i** (entry 8) furnished the *trans*-*unlike*-configured main product *tu*-**3i** in acceptable yield; the use of the diastereomeric **2h** (entry 7) gave no product. The application of 2-ethoxy-1,3-oxazolidines such as **2g** (entries 5 and 6) or **2j** (entries 9 and 10) gave mixtures of similar amounts of the *cis*-*unlike*- and *cis*-*like*-isomers. The *cis*- and *trans*-epimers of **2g** and also of **2j** were subjected separately to the reaction with the enolate **4b** (Table 5, entries 5 and 6, 9 and 10); as expected, the C-2 configuration showed no evident influence on the observed diastereomeric ratios. Good *cis*-*unlike*-selectivities were obtained with the 2-ethoxy-1,3-oxazolidines **2c**, **2f** and **2k** (entries 1, 4 and 11). The reagents **2c** and **2k** appeared to be most valuable because they provided the desired products **3c** and **3k** in good yields and selectivities.

Some experiments (Table 5, entries 3–5, 8, 9, 11) furnished up to 10% of the isomeric by-products **6**, which were identified easily by the <sup>1</sup>H NMR signal of the 2'-H (doublet instead of singlet). An isomerization of the titanium enolate **4b** might be responsible for this result (Scheme 4), since the used silyl enol ether **1b** was essentially isomerically pure<sup>13</sup> (GC analysis: >97:3). Perhaps, traces of a proton source in some of the 2-ethoxy-1,3-oxazolidines are catalyzing this process.

The 2-ethoxy-1,3-oxazolidine **2k** was allowed to react with three other titanium enolates obtained from the known<sup>4</sup> silyl enol ethers **1c–e** (Table 5, entries 12–14). The *cis*-*unlike*-configured main products *cu*-**3** were produced in acceptable

<sup>13</sup> Silyl enol ethers **1b** and **1e** were purified by spaltrohr distillation.



Scheme 4.

to excellent selectivities and moderate to good yields. In particular the cyclopentanone derivative *cis*-**3n** was formed with high selectivity.

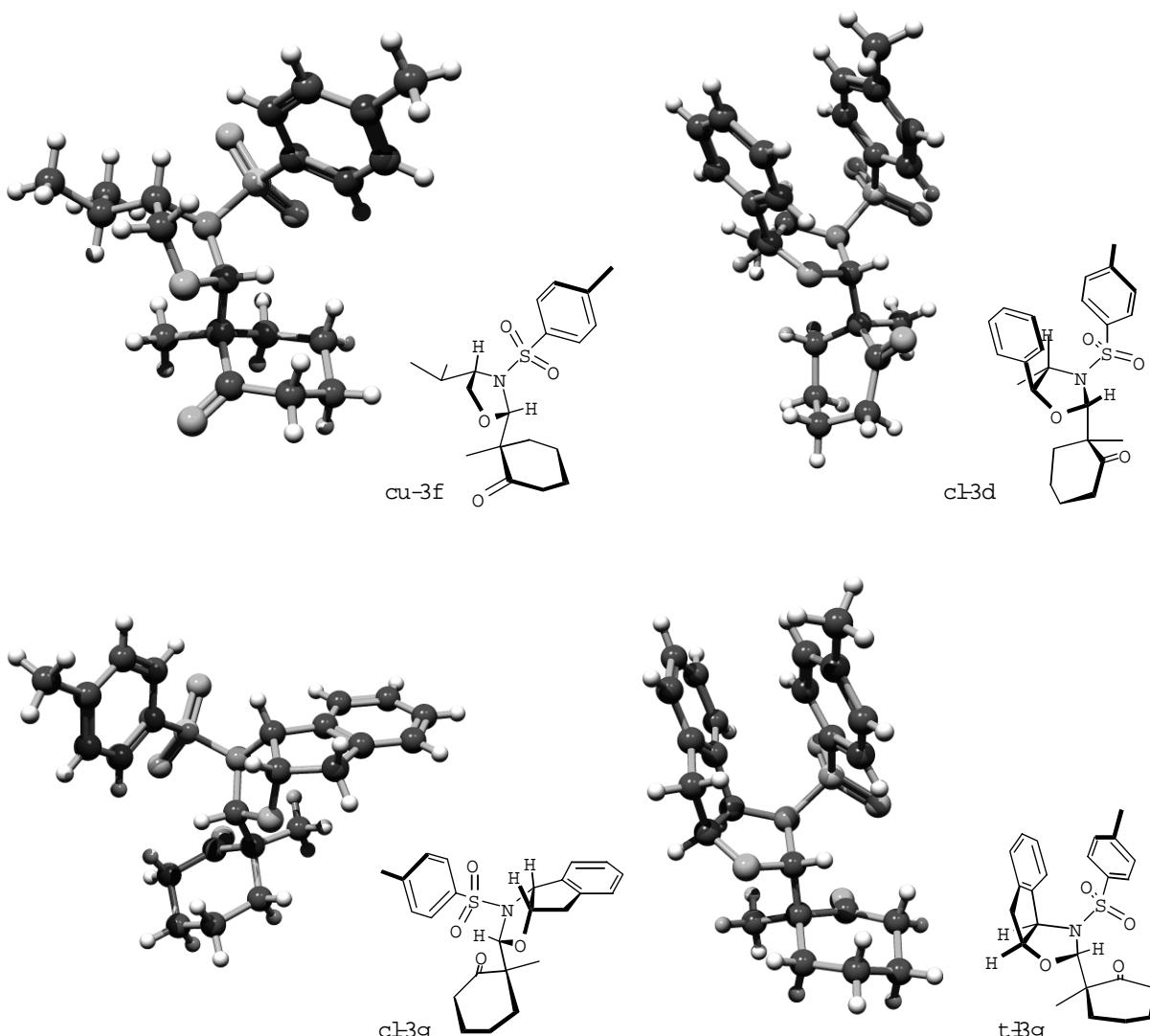
The configurations of the synthesized 1,3-oxazolidines **3** were proven by X-ray analyses<sup>12</sup> (Figs. 3–5) in many cases; some more configurations could be determined by drawing conclusions from the NMR data, selected shifts are presented in Table 6. In accordance with the X-ray

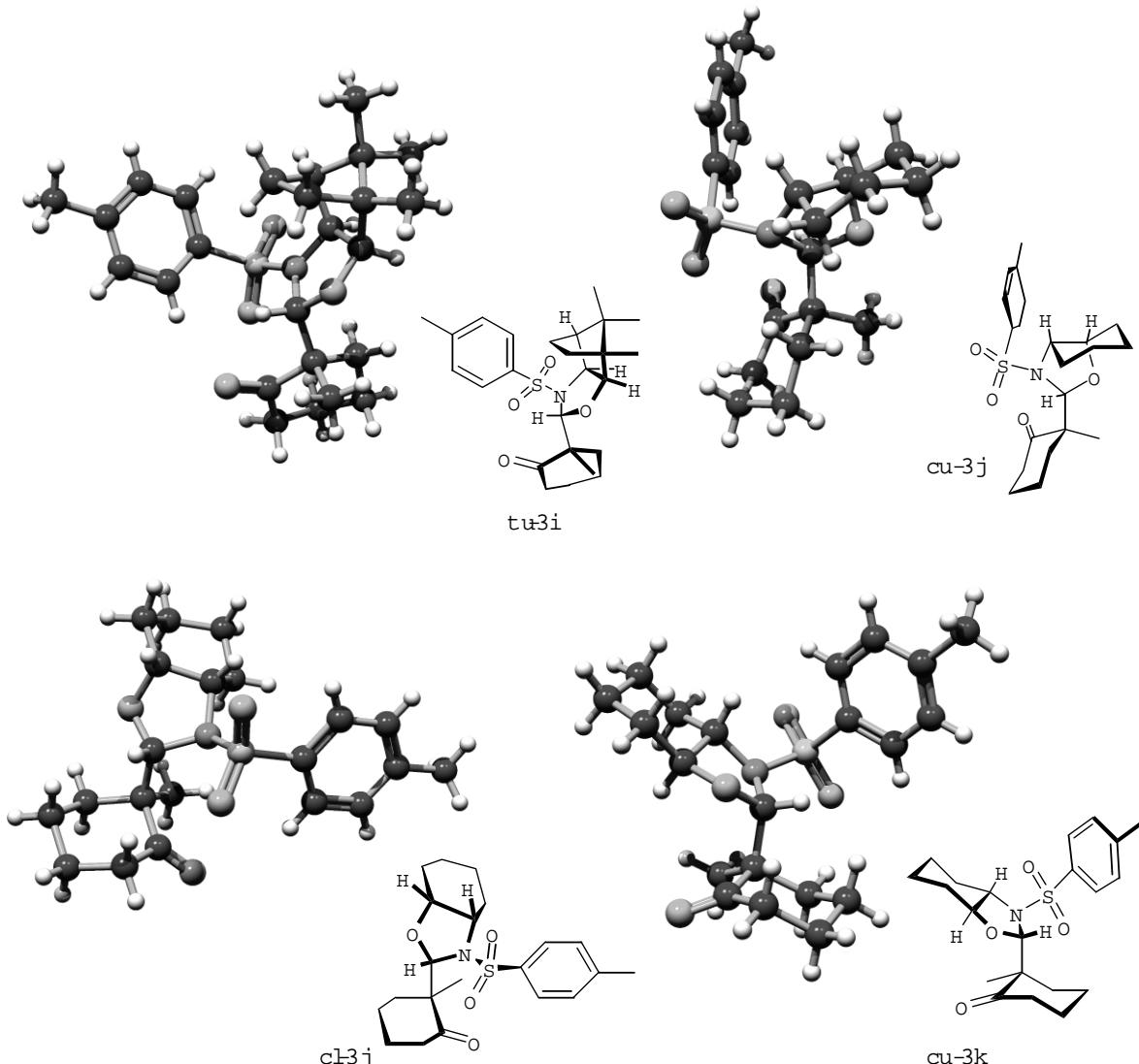
analyses the signals of the 2-H and C-2 of the *cu*-configured products *cu*-**3** are shifted more to high field than the signals of the other diastereomers.

Some relationships between the structures of the 2-ethoxy-1,3-oxazolidines **2** and the observed diastereoselectivities of their reaction with the titanium enolate **4b** can be recognized: To obtain a good selectivity in favour of the *cis*-*unlike*-product *cu*-**3**, 2-ethoxy-1,3-oxazolidines with a

**Table 5.** Results of the reactions of silyl enol ethers **1** and 2-ethoxy-3-sulfonyl-1,3-oxazolidines **2** (method C)

Entry	Educts <sup>a</sup>	Product	Yield (%) <sup>b</sup>	dr <sup>c</sup>	By-product <sup>d</sup>
1	<b>1b</b> <sup>4a</sup> + <b>2c</b> (99:1)	<b>3c</b> <sup>3k</sup>	79	81 ( <i>cu</i> ):19 ( <i>cl</i> )	<b>6c</b> , 0%
2	<b>1b</b> + <b>2d</b> (90:10) <sup>e</sup>	<b>3e</b>	44	74 ( <i>cu</i> ):26 ( <i>cl</i> )	<b>6e</b> , 0%
3	<b>1b</b> + <b>2e</b> (95:5)	<b>3f</b>	54	66 ( <i>cu</i> ):34 ( <i>cl</i> )	<b>6f</b> , 8%, dr=50:50
4	<b>1b</b> + <b>2f</b> (>97:3)	<i>ent</i> - <b>3d</b> <sup>1b</sup>	50	92 ( <i>cu</i> ):8 ( <i>cl</i> )	<b>6d</b> , 8%, dr=100:0
5	<b>1b</b> + <b>2g</b> (97:3)	<b>3g</b>	74	35 ( <i>cl</i> ):34 ( <i>cu</i> ):25 ( <i>tl</i> ):6 ( <i>tu</i> )	<b>6g</b> , 10%, dr=60:40
6	<b>1b</b> + <b>2g</b> (6:94)	<b>3g</b>	— <sup>f</sup>	35 ( <i>cl</i> ):38 ( <i>cu</i> ):24 ( <i>tl</i> ):3 ( <i>tu</i> )	<b>6g</b> , — <sup>f</sup>
7	<b>1b</b> + <b>2h</b> (100:0) <sup>g</sup>	<b>3h</b>	0	—	<b>6h</b> , 0%
8	<b>1b</b> + <b>2i</b> (86:14) <sup>h</sup>	<b>3i</b>	57	75 ( <i>tu</i> ):13:5:7	<b>6i</b> , 7%, dr=93:7
9	<b>1b</b> + <b>2j</b> (100:0)	<b>3j</b>	38	43 ( <i>cl</i> ):36 ( <i>cu</i> ):21 ( <i>t</i> )	<b>6j</b> , 1%, dr=75:25
10	<b>1b</b> + <b>2j</b> (4:96)	<b>3j</b>	— <sup>f</sup>	45 ( <i>cl</i> ):36 ( <i>cu</i> ):19 ( <i>t</i> )	<b>6j</b> , — <sup>f</sup>
11	<b>1b</b> + <b>2k</b> (100:0)	<b>3k</b>	74	89 ( <i>cu</i> ):11	<b>6k</b> , 2%, dr=100:0
12	<b>1c</b> <sup>4b</sup> + <b>2k</b> (100:0)	<b>3l</b>	47	78 ( <i>cu</i> ):13 ( <i>cl</i> ):9 ( <i>t</i> )	—
13	<b>1d</b> <sup>4c</sup> + <b>2k</b> (100:0)	<b>3m</b>	68	74 ( <i>cu</i> ):19:4:3	—
14	<b>1e</b> <sup>4d</sup> + <b>2k</b> (100:0)	<b>3n</b>	59	98 ( <i>cu</i> ):2	<b>6n</b> , 0%

<sup>a</sup> The diastereomeric ratios of the 2-ethoxy-1,3-oxazolidines **2** are given in parenthesis.<sup>b</sup> All yields were determined after flash column chromatography.<sup>c</sup> All diastereomeric ratios were determined by <sup>1</sup>H NMR spectroscopy of the crude products. Bold specifications of configurations were determined by X-ray analyses.<sup>d</sup> All yields and diastereomeric ratios were determined after flash column chromatography.<sup>e</sup> Contained 8% of triethyl orthoformate.<sup>f</sup> Not determined.<sup>g</sup> Contained 12% of triethyl orthoformate.<sup>h</sup> Contained 4% of 2-ethoxy-1,3-oxazolidine **2h**.**Figure 3.** Molecular structures of 3-sulfonyl-1,3-oxazolidines *cu*-**3f**, *cl*-**3d**, *cl*-**3g** and *tl*-**3g**.<sup>12</sup>



**Figure 4.** Molecular structures of 3-sulfonyl-1,3-oxazolidines *tu*-3*i*, *cu*-3*j*, *cl*-3*j* and *cu*-3*k*<sup>12</sup> (X-ray analysis for *cu*-3*j*, *cl*-3*j* and *cu*-3*k* were obtained from a racemic sample).

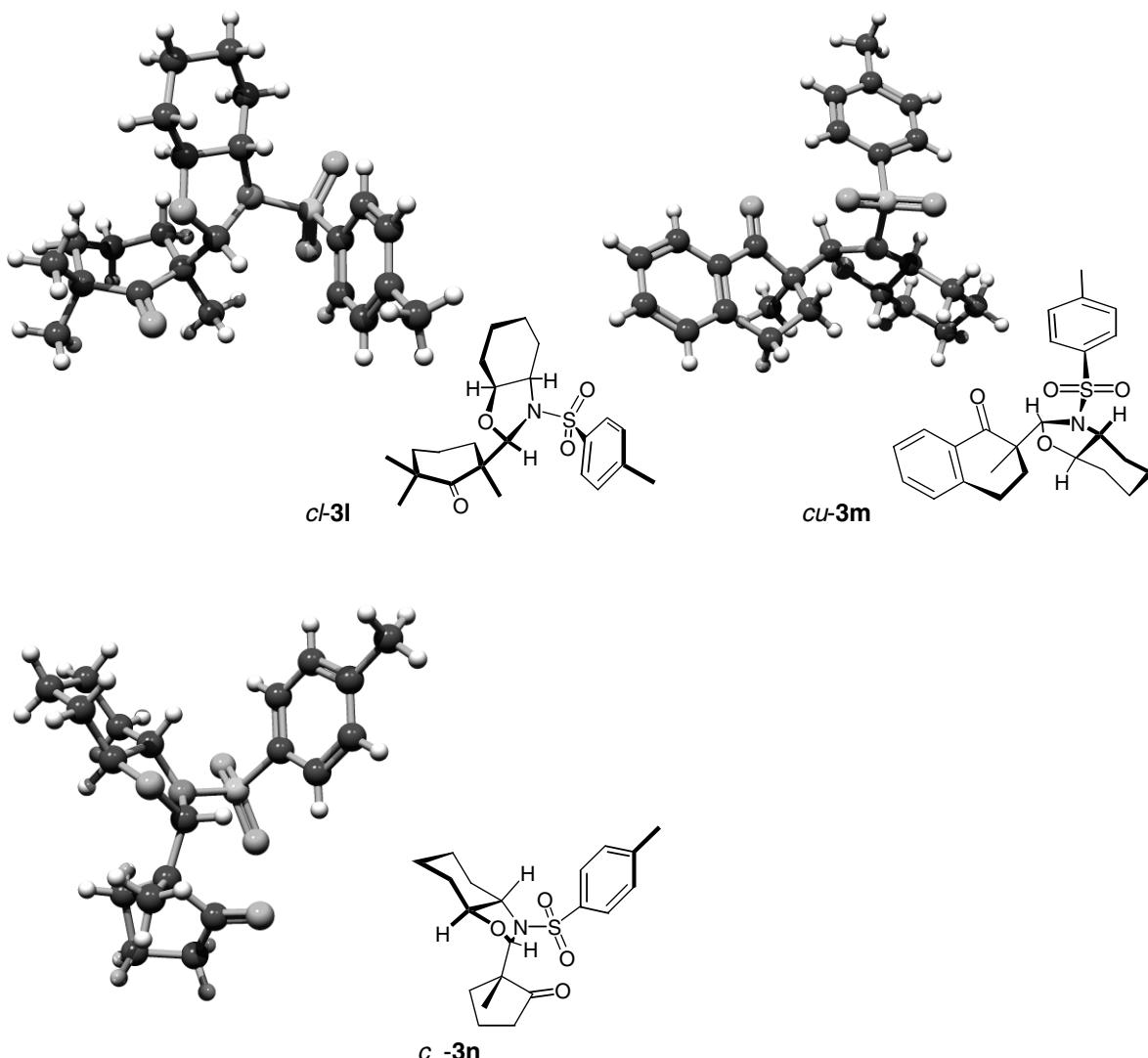
small group in the 4 position of the 1,3-oxazolidine ring (**2c**) or with two *trans*-configured groups in the 4 and 5 position (**2f**, **2k**) should be used. Two *cis*-orientated groups in the 4 and 5 position (**2g**, **2j**) lead to the formation of *cis-like*-product *cl*-3 at the expense of *cis-unlike*-product *cu*-3 and furthermore, *trans*-products *t*-3 are detected or even become the main diastereomer (**2i**). A large group in 4 position (**2e**) or a large sulfonyl group (**2d**) support the formation of the *cis-like*-configured products *cl*-3, as well.

The observed *cis* selectivity in most reactions can be rationalized as an addition of the enolate *trans* to the bulky sulfonyl group, which is in a *trans* position relative to the adjacent C-4 substituent.<sup>13</sup> The formation of *trans*-products *t*-3 instead of the generally preferred *cis*-products *c*-3 may be consistent with the strong steric shielding of the *cis*-side in 2-ethoxy-1,3-oxazolidines like **2i** (and also **2g** and **2j**).

The *like* or *unlike* selectivity for the formation of the *cis*-isomers *c*-3 is more difficult to understand. To explain the different results of the reactions of 2-alkoxy-1,3-oxa-

zolidines **2** either with the silyl enol ether **1b** or the titanium enolate **4b** we suggest a cyclic transition state in the latter case. The titanium atom of the enolate may coordinate to the 1,3-oxazolidine oxygen atom or to the sulfonyl group (Fig. 6). Perhaps a coordination to the 1,3-oxazolidine oxygen atom (TS **A** and TS **B**) increases the electrophilicity of the cation and therefore promotes the reaction with the nucleophilic enolate, no matter which coordination may be thermodynamically favoured. We think that in this case TS **B** should be disfavoured because of steric interactions between the 1,3-oxazolidine ring and the cyclohexyl ring and this could be the reason for the *cis-unlike*-configuration of the main products in most cases.

A closer view at some of the possible geometries of the transition states TS **A** and TS **B** is given in Fig. 7. The transition states TS **A**<sup>3</sup> and TS **B**<sup>3</sup> appear to be disfavoured because of the titanium atom (and its ligands) placed at the top of a boat-like geometry; this creates a strong transannular strain. Two chair-like (TS **A**<sup>1</sup> and TS **B**<sup>2</sup>) and two boat-like geometries (TS **A**<sup>2</sup> and TS **B**<sup>1</sup>) can be regarded for explanations. Like in the corresponding aldol reactions<sup>6b</sup> of



**Figure 5.** Molecular structures of 3-sulfonyl-1,3-oxazolidines *cl*-3l, *cu*-3m and *cu*-3n.<sup>12</sup>

**Table 6.** Selected NMR data of the 3-sulfonyl-1,3-oxazolidines **3**

Entry	Compound	<i>cu</i>		<i>cl</i>		<i>tu</i>		<i>tl</i>	
		2-H	C-2	2-H	C-2	2-H	C-2	2-H	C-2
1	<b>3c</b>	<b>5.36<sup>a</sup></b>	<b>93.6</b>	5.69	94.3				
2	<b>3e</b>	5.97	91.4	6.01	92.5				
3	<b>3f</b>	<b>5.36</b>	<b>93.9</b>	5.66	94.6				
4	<i>ent</i> - <b>3d</b>	5.88	93.8	<b>6.11</b>	<b>95.5</b>				
5	<b>3g</b>	5.67	94.5	<b>5.89</b>	<b>95.4</b>	5.99	95.7	<b>6.36</b>	<b>99.0</b>
6	<b>3i</b>	5.71 <sup>b,c</sup>	— <sup>d</sup>	6.03 <sup>b,c</sup>	94.6 <sup>c,e</sup>	<b>5.96<sup>b</sup></b>	<b>97.7<sup>c</sup></b>	6.36 <sup>b,c</sup>	99.7 <sup>c,e</sup>
7	<b>3j</b>	<b>5.34</b>	<b>93.0</b>	<b>5.65</b>	<b>92.9</b>	5.78 <sup>c</sup>	90.8 <sup>c</sup>		
8	<b>3k</b>	<b>5.42</b>	<b>92.3</b>	5.78	93.1				
9	<b>3l</b>	5.40	93.4	<b>5.44</b>	<b>96.4</b>	6.07 <sup>c</sup>	95.6 <sup>c</sup>		
10	<b>3m</b>	<b>5.56</b>	<b>93.2</b>	5.61 <sup>c</sup>	93.7 <sup>c</sup>	5.80 <sup>c</sup>	— <sup>d</sup>	6.18 <sup>c</sup>	— <sup>d</sup>
11	<b>3n</b>	<b>5.01</b>	<b>93.2</b>	5.13	— <sup>d</sup>				

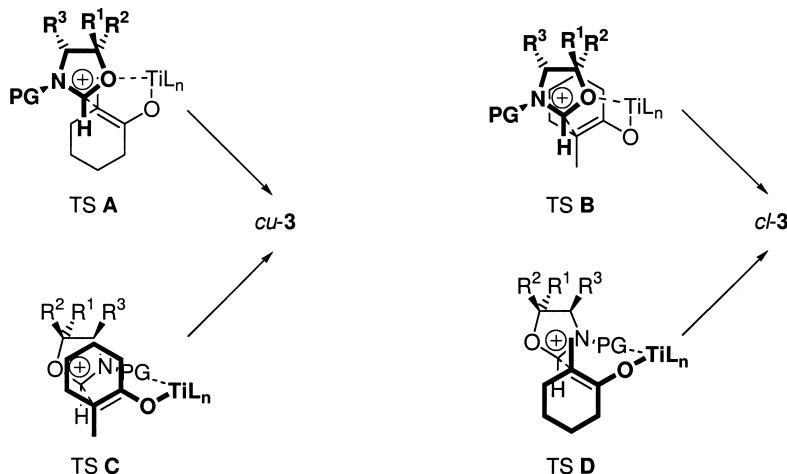
<sup>a</sup> Bold shifts indicate that the corresponding configuration was determined by X-ray analyses.

<sup>b</sup> 4-H is the corresponding proton.

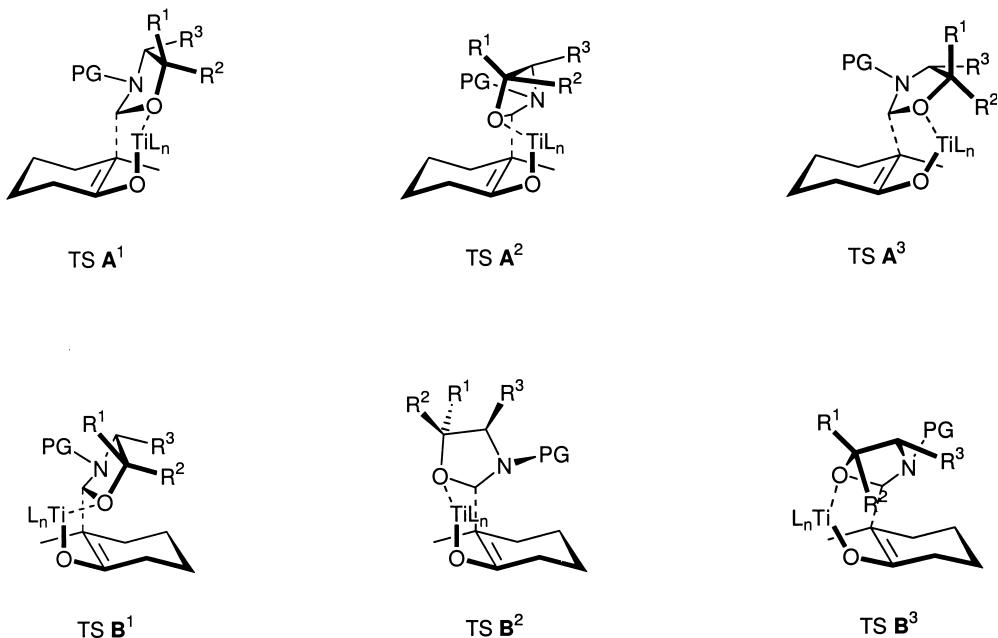
<sup>c</sup> The configuration of this product is uncertain.

<sup>d</sup> Not detected.

<sup>e</sup> C-4 is the corresponding carbon atom.



**Figure 6.** Proposed transition states **A**, **B**, **C** and **D**.



**Figure 7.** Proposed geometries of the transition states **A** and **B**.

trichlorotitanium enolates, only boat-like transition state geometries are able to explain the observed results. In the suggested model steric interactions between the titanium fragment and the groups  $R^2$  and  $R^3$  as well as repulsions between the sulfonyl group and the cyclohexyl ring disfavour the *cis-unlike*-configuration (TS  $A^2$ ); interactions between the  $R^1$  group and the titanium atom and its ligands disfavour the formation of *cis-like*-configured products (TS  $B^1$ ). The chair-like geometries of TS  $A^1$  and TS  $B^2$  have the opposite ‘configuration’ at the 1,3-oxazolidine oxygen atom and would predict just the opposite influence of the groups  $R^1$ ,  $R^2$  and  $R^3$ .

### 3. Conclusions

In the present study, the reaction of chiral 2-ethoxy-3-sulfonyl-1,3-oxazolidines **2** and prochiral titanium enolates **4** has been investigated. It was found that the reaction of

reagents like **2c** or **2k**, derived from the *N*-tosyl derivatives of 2-aminobutanol and *trans*-2-aminocyclohexanol with trichlorotitanium enolates in dichloromethane represents a useful method to install quaternary  $\alpha$ -carbonyl stereocentres (after deprotection of the chirally masked formyl group). The *cis-unlike*-configured products *cu-3* are obtained in acceptable to good yields and selectivities. Some details of the possible mechanism of this reaction are described and the observed selectivities are interpreted. In summary, this new method is a valuable supplement to the known asymmetric formylation reactions and allows the stereo-selective formation of synthetically interesting compounds.

### 4. Experimental

#### 4.1. General

All reactions were performed under Ar in flame-dried glass-





[89.8] (d, C-2); 111.2 [112.3] (d, C-4); 128.6, 128.9 [127.6, 129.4] (2d, C-2', C-3', C-5', C-6'); 136.3 (s, C-4'); 143.4 (s, C-1'). IR (KBr): 2965 (m); 2885 (m); 1730 (w); 1605 (w); 1360 (s); 1180 (s); 1105 (s); 1040 (s); 685 (s). EI-MS (70 eV): 379 (4, M<sup>+</sup>); 350 (100); 334 (85); 224 (14); 196 (28); 155 (32); 150 (50); 135 (28); 123 (51); 109 (25); 91 (77). Anal. calcd for C<sub>20</sub>H<sub>29</sub>NO<sub>4</sub>S (379.52): C 63.30, H 7.70, N 3.69; found: C 62.93, H 7.66, N 3.72.

**4.2.8. (3a*R*,7a*S*)-2-Ethoxy-3-(4-methylbenzenesulfonyl)-perhydro-1,3-benzoxazazole (**2j**).** According to GP A, sulfonamide **5j**<sup>10</sup> (328 mg, 1.22 mmol) was treated with triethyl orthoformate (10 mL, 8.9 g, 60 mmol) to provide the crude product **2j** (*a/b*=54:46). FCC (125 cm<sup>3</sup> SiO<sub>2</sub>, Et<sub>2</sub>O/PE=1:5→1:1) afforded *a*-**2j** (158 mg, 0.486 mmol, 40%) as a colourless oil and a mixture of *b*-**2j** and *a*-**2j** (148 mg, 0.455 mmol, 37%, dr=96:4) as a white wax. *a*-**2j**: R<sub>F</sub> (SiO<sub>2</sub>, Et<sub>2</sub>O/PE=1:1)=0.70. [α]<sub>D</sub><sup>20</sup>=−29.0 (*c*=0.42, CH<sub>2</sub>Cl<sub>2</sub>). <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): 1.02–1.18, 1.30–1.71, 1.96–2.06, 2.17–2.28 (m, m, ‘d’, m, 8H, 5-H<sub>2</sub>, 6-H<sub>2</sub>, 7-H<sub>2</sub>, 8-H<sub>2</sub>); 1.13 (t, *J*=7.5 Hz, 3H, 2"-H<sub>3</sub>); 2.39 (s, 3H, 4'-CH<sub>3</sub>); 3.46–3.66, 4.15–4.20 (2m, 3H, 1H, 5-H, 9-H, 1"-H<sub>2</sub>); 5.89 (s, 1H, 2-H); 7.23 (d, *J*=7.8 Hz, 2H, 3'-H, 5'-H); 7.80 (‘d’, *J*=8.4 Hz, 2H, 2'-H, 5'-H). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): 14.8 (q, C-2"); 19.5, 22.2 (2t, C-6, C-7); 21.5 (q, 4'-CH<sub>3</sub>); 26.8, 29.8 (2t, C-5, C-8); 56.7 (d, C-4); 62.6 (t, C-1"); 73.7 (d, C-9); 105.8 (d, C-2); 127.7 (d, C-2', C-6'); 129.1 (d, C-3', C-5'); 138.6 (s, C-4'); 143.0 (s, C-1'). IR (KBr): 2940 (m); 2870 (w); 1605 (w); 1455 (w); 1355 (s); 1165 (s); 1070 (s); 950 (w); 815(w). EI-MS (70 eV): 325 (2.6, M<sup>+</sup>); 280 (100); 210 (15); 200 (18); 170 (11); 155 (55); 96 (21); 91 (83). Anal. calcd for C<sub>16</sub>H<sub>23</sub>NO<sub>4</sub>S (325.43): C 59.05, H 7.12, N 4.30; found: C 59.08, H 7.16, N 4.26.

*b*-**2j/a**-**2j**=96:4: R<sub>F</sub> (SiO<sub>2</sub>, Et<sub>2</sub>O/PE=1:1)=0.67. Mp (Et<sub>2</sub>O/PE)>90°C. [α]<sub>D</sub><sup>20</sup>=+10.3 (*c*=0.74, CH<sub>2</sub>Cl<sub>2</sub>). <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): 1.02–1.17, 1.32–1.85, 1.95–2.04 (3m, 8H, 5-H<sub>2</sub>, 6-H<sub>2</sub>, 7-H<sub>2</sub>, 8-H<sub>2</sub>); 1.21 (t, *J*=6.9 Hz, 3H, 2"-H<sub>3</sub>); 2.41 (s, 3H, 4'-CH<sub>3</sub>); 3.55–3.81 (m, 4H, 5-H, 9-H, 1"-H<sub>2</sub>); 5.97 [5.89] (s, 1H, 2-H); 7.28 (d, *J*=7.8 Hz, 2H, 3'-H, 5'-H); 7.75 (‘d’, *J*=8.4 Hz, 2H, 2'-H, 5'-H). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): 15.0 (q, C-2"); 19.7, 22.5 (2t, C-6, C-7); 21.5 (q, 4'-CH<sub>3</sub>); 27.7, 28.7 (2t, C-5, C-8); 56.3 (d, C-4); 62.3 (t, C-1"); 75.9 (d, C-9); 107.7 (d, C-2); 127.4 (d, C-2', C-6'); 129.7 (d, C-3', C-5'); 137.3 (s, C-4'); 143.7 (s, C-1'). IR (KBr): 2950 (s); 2860 (m); 1730 (m); 1600 (m); 1350 (s); 1165 (s); 1080 (s); 980 (s); 680(s); 600(s). EI-MS (70 eV): 325 (1.8, M<sup>+</sup>); 280 (95); 210 (18); 200 (17); 170 (13); 155 (59); 96 (27); 91 (100). Anal. calcd for C<sub>16</sub>H<sub>23</sub>NO<sub>4</sub>S (325.43): C 59.05, H 7.12, N 4.30; found: C 58.65, H 6.98, N 4.45.

**4.2.9. (3a*S*,7a*S*)-2-Ethoxy-3-(4-methylbenzenesulfonyl)-perhydro-1,3-benzoxazazole (**2k**).** According to GP A, sulfonamide **5k**<sup>11</sup> (1.16 g, 4.31 mmol) was treated with triethyl orthoformate (15 mL, 13 g, 90 mmol) to provide the crude product **2k** (*a/b*=73:27). FCC (280 cm<sup>3</sup> SiO<sub>2</sub>, Et<sub>2</sub>O/PE=1:3→1:1) afforded *a*-**2k** (1.26 g, 3.87 mmol, 90%) as a colourless oil and only traces of *b*-**2k** (20 mg, <sup>1</sup>H NMR: 5.02 (s, 1H, 2-H), mixture with other compounds) as a colourless oil. *a*-**2k**: R<sub>F</sub> (SiO<sub>2</sub>, Et<sub>2</sub>O/PE=1:1)=0.45. [α]<sub>D</sub><sup>20</sup>=−2.8 (*c*=0.50, CH<sub>2</sub>Cl<sub>2</sub>). <sup>1</sup>H NMR (300 MHz,

CDCl<sub>3</sub>): 1.13–1.45, 1.67–1.74, 1.99–2.09, 2.23–2.35 (4m, 4H, 2H, 1H, 1H, 5-H<sub>2</sub>, 6-H<sub>2</sub>, 7-H<sub>2</sub>, 8-H<sub>2</sub>); 1.22 (t, *J*=6.6 Hz, 3H, 2"-H<sub>3</sub>); 2.40 (s, 3H, 4'-CH<sub>3</sub>); 2.71, 3.60–3.82 (ddd, m, *J*=3.3, 9.9, 11.1 Hz, 1H, 3H, 4-H, 9-H, 1"-H<sub>2</sub>); 5.85 (s, 1H, 2-H); 7.28 (d, *J*=8.7 Hz, 2H, 3'-H, 5'-H); 7.73 (d, *J*=8.4 Hz, 2H, 2'-H, 6'-H). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): 15.0 (q, C-1"); 21.5 (4'-CH<sub>3</sub>); 23.4, 23.8 (2t, C-6, C-7); 28.8, 29.0 (2t, C-5, C-8); 62.7 (t, C-2"); 64.4 (d, C-4); 80.3 (d, C-9); 107.1 (d, C-2); 127.5 (d, C-2', C-6'); 129.6 (d, C-3', C-5'); 136.1 (s, C-4'); 143.7 (s, C-1'). IR (KBr): 2965 (m); 2880 (m); 1605 (w); 1445 (w); 1360 (s); 1170 (s); 1140 (s); 1085 (s); 1045 (s); 830 (m); 660 (s); 590 (s). EI-MS (70 eV): 325 (3.3, M<sup>+</sup>); 280 (35); 210 (27); 155 (44); 139 (10); 96 (60); 91 (100). Anal. calcd for C<sub>16</sub>H<sub>23</sub>NO<sub>4</sub>S (325.43): C 59.05, H 7.12, N 4.30; found: C 58.97, H 7.28, N 4.13.

#### 4.3. Syntheses of 2-methyl-2-(3-sulfonyl-1,3-oxazolidin-2-yl)cycloalkanones **3**

**4.3.1. General procedure B (GP B).** To a cold (−78°C) solution of the silyl enol ether **1** (1.00 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 mL), TiCl<sub>4</sub> (0.11 mL, 0.19 g, 1.0 mmol) was added dropwise. After stirring for 15 min at −78°C, the reaction mixture was allowed to warm to room temperature and stirred for a further 45 min. The deep-red solution was then cooled to −78°C once more, whereupon a solution of the 2-ethoxy-1,3-oxazolidine **2** (1.00 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (4.0 mL) was slowly added. The resulting mixture was allowed to warm to room temperature over a period of 15 h. The reaction was then stopped by the addition of satd. aqueous NaHCO<sub>3</sub> (20 mL) and the mixture was given to satd. aqueous NaCl (50 mL) and CH<sub>2</sub>Cl<sub>2</sub> (50 mL). The aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3×50 mL) and the combined organic layers were stirred with K<sub>2</sub>CO<sub>3</sub> (2.0 g) for 15 min. The suspension was filtered (Celite) and the solvent was removed in vacuo to leave the crude 2-(1,3-oxazolidin-2-yl)-cycloalkanone **3**.

#### 4.3.2. 2-[(*2R,4R*)-4-Ethyl-3-(4-methylbenzenesulfonyl)-1,3-oxazolidin-2-yl]-2-methylcyclohexanone (**3c**).<sup>3k</sup>

Method A: A 2.2 M solution of ZnCl<sub>2</sub>·Et<sub>2</sub>O in CH<sub>2</sub>Cl<sub>2</sub> (7.3 mL, 16 mmol) was added to a solution of silyl enol ether **1b**<sup>4a</sup> (2.95 g, 16.0 mmol) and 2-ethoxy-1,3-oxazolidine *cis*-**2c**<sup>3k</sup> (4.79 g, 16.0 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (75 mL) at 0°C. After stirring at 0°C for 1 h, satd. aqueous NaCl (75 mL) was added and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3×50 mL). The combined organic layers were dried with MgSO<sub>4</sub> (5.0 g) and the solvent was removed in vacuo. Purification of the crude product **3c** (*cl/cu*=51:49) by FCC (825 cm<sup>3</sup> SiO<sub>2</sub>, Et<sub>2</sub>O/PE=1:5→1:2.5) provided *cis*-**3c**<sup>3k</sup> (2.33 g, 6.37 mmol, 40%) and *cl*-**3c**<sup>3k</sup> (2.49 g, 6.81 mmol, 43%) as white solids.

Method B: A 1.6 M solution of MeLi in Et<sub>2</sub>O (0.63 mL, 1.0 mmol) was added to a solution of silyl enol ether **1b**<sup>4a</sup> (184 mg, 1.00 mmol) in toluene (5.0 mL). The mixture was heated to 80°C for 1 h and then cooled to −78°C. A 1.0 M solution of TiCl(O*i*Pr)<sub>3</sub> in toluene (1.0 mL, 1.0 mmol) was added and the resulting red suspension was stirred at room temperature for 30 min. After cooling to −78°C once more, a solution of 2-ethoxy-1,3-oxazolidine *cis*-**2c**<sup>3k</sup> (299 mg, dr=98:2) in toluene (1.0 mL) and then BF<sub>3</sub>·Et<sub>2</sub>O (0.25 mL, 0.28 g, 2.0 mmol) were added and the reaction













( $0.688 \leq T \leq 0.793$ ),  $Z=4$ , monoclinic, space group  $P2_1/c$  (No. 14),  $\lambda=1.54178 \text{ \AA}$ ,  $T=223 \text{ K}$ ,  $\omega/2\theta$  scans, 4304 reflections collected ( $\pm h, +k, -l$ ),  $[(\sin \theta)/\lambda]=0.62 \text{ \AA}^{-1}$ , 4174 independent ( $R_{\text{int}}=0.021$ ) and 3128 observed reflections [ $I \geq 2\sigma(I)$ ], 246 refined parameters,  $R=0.041$ ,  $wR^2=0.106$ , max. residual electron density 0.21 ( $-0.45$ )  $e \text{ \AA}^{-3}$ , hydrogens calculated and refined as riding atoms.

**4.4.10. Crystal data for cl-3l.** Formula  $C_{23}H_{33}NO_4S$ ,  $M=419.56$ , colourless crystal  $0.25 \times 0.20 \times 0.10 \text{ mm}^3$ ,  $a=10.087(3)$ ,  $b=11.272(4)$ ,  $c=19.863(3) \text{ \AA}$ ,  $V=2258.4(8) \text{ \AA}^3$ ,  $\rho_{\text{calc}}=1.234 \text{ g cm}^{-3}$ ,  $\mu=14.96 \text{ cm}^{-1}$ , empirical absorption correction via  $\psi$  scan data ( $0.706 \leq T \leq 0.865$ ),  $Z=4$ , orthorhombic, space group  $P2_12_12_1$  (No. 19),  $\lambda=1.54178 \text{ \AA}$ ,  $T=223 \text{ K}$ ,  $\omega/2\theta$  scans, 2607 reflections collected ( $+h, +k, +l$ ),  $[(\sin \theta)/\lambda]=0.62 \text{ \AA}^{-1}$ , 2607 independent and 1791 observed reflections [ $I \geq 2\sigma(I)$ ], 267 refined parameters,  $R=0.047$ ,  $wR^2=0.111$ , max. residual electron density 0.20 ( $-0.26$ )  $e \text{ \AA}^{-3}$ , Flack parameter 0.07(4), hydrogens calculated and refined as riding atoms.

**4.4.11. Crystal data for cu-3m.** Formula  $C_{25}H_{29}NO_4S$ ,  $M=439.55$ , colourless crystal  $0.45 \times 0.20 \times 0.10 \text{ mm}^3$ ,  $a=8.010(1)$ ,  $b=8.542(3)$ ,  $c=32.041(3) \text{ \AA}$ ,  $V=2192.3(8) \text{ \AA}^3$ ,  $\rho_{\text{calc}}=1.332 \text{ g cm}^{-3}$ ,  $\mu=15.74 \text{ cm}^{-1}$ , empirical absorption correction via  $\psi$  scan data ( $0.538 \leq T \leq 0.859$ ),  $Z=4$ , orthorhombic, space group  $P2_12_12_1$  (No. 19),  $\lambda=1.54178 \text{ \AA}$ ,  $T=223 \text{ K}$ ,  $\omega/2\theta$  scans, 2575 reflections collected ( $-h, -k, -l$ ),  $[(\sin \theta)/\lambda]=0.62 \text{ \AA}^{-1}$ , 2575 independent and 2277 observed reflections [ $I \geq 2\sigma(I)$ ], 283 refined parameters,  $R=0.053$ ,  $wR^2=0.134$ , max. residual electron density 0.35 ( $-0.44$ )  $e \text{ \AA}^{-3}$ , Flack parameter  $-0.05(4)$ , hydrogens calculated and refined as riding atoms.

**4.4.12. Crystal data for cu-3n.** Formula  $C_{20}H_{27}NO_4S$ ,  $M=377.49$ , colourless crystal  $0.35 \times 0.15 \times 0.10 \text{ mm}^3$ ,  $a=9.834(2)$ ,  $b=12.360(2)$ ,  $c=16.166(3) \text{ \AA}$ ,  $V=1964.9(6) \text{ \AA}^3$ ,  $\rho_{\text{calc}}=1.276 \text{ g cm}^{-3}$ ,  $\mu=16.64 \text{ cm}^{-1}$ , empirical absorption correction via  $\psi$  scan data ( $0.594 \leq T \leq 0.851$ ),  $Z=4$ , orthorhombic, space group  $P2_12_12_1$  (No. 19),  $\lambda=1.54178 \text{ \AA}$ ,  $T=223 \text{ K}$ ,  $\omega/2\theta$  scans, 2290 reflections collected ( $-h, +k, +l$ ),  $[(\sin \theta)/\lambda]=0.62 \text{ \AA}^{-1}$ , 2290 independent and 2150 observed reflections [ $I \geq 2\sigma(I)$ ], 237 refined parameters,  $R=0.033$ ,  $wR^2=0.092$ , max. residual electron density 0.24 ( $-0.26$ )  $e \text{ \AA}^{-3}$ , Flack parameter 0.04(2), hydrogens calculated and refined as riding atoms.

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1997), graphics MOPICT 3.2 (Brüggemann, M., Universität Münster, 2001). Crystallographic data (excluding structure factors) for structures **cis-2c**, **cu-3f**, **cl-3d**, **cl-3g**, **tl-3g**, **tu-3i**, **cu-3j**, **cl-3j**, **cu-3k**, **cl-3l**, **cu-3m** and **cu-3n** reported in this paper have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication numbers CCDC-166227, CCDC-166228, CCDC-166229, CCDC-166230, CCDC-166231, CCDC-166232, CCDC-166233,

CCDC-166234, CCDC-166235, CCDC-166236, CCDC-166237 and CCDC-166238. Copies of the data can be obtained, free of charge, on application to the CCDC, 12 Union Road, Cambridge CB2 1EZ, UK (fax: +44-(0)1223-336033; e-mail: deposit@ccdc.cam.ac.uk).

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